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***Schaum's Outline of Thermodynamics for Engineers,
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Thermodynamics for Engineers, 2nd Edition
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Engineers, 2nd Edition Technical Thermodynamics for
Engineers Thermodynamics for Engineers Non-
Equilibrium Thermodynamics for Engineers Modern
Engineering Thermodynamics Thermodynamics for
Engineers Thermodynamics for the Practicing Engineer
Non-equilibrium Thermodynamics For Engineers
(Second Edition) Heat Engineering Thermodynamic
Approaches in Engineering Systems Statistical
Thermodynamics Advanced Thermodynamics
Engineering, Second Edition Schaum's Outline of
Thermodynamics for Engineers, 3rd Edition Applied
Thermodynamics for Engineering Technologists
Thermodynamics for Chemists, Physicists and
Engineers Engineering Thermodynamics Chemical
Energy and Exergy Thermodynamics for Engineers
Thermodynamics Energy Systems Thermodynamics
with Chemical Engineering Applications Elements of
Irreversible Thermodynamics for Engineers Molecular
Engineering Thermodynamics Fundamentals of***

**Engineering Thermodynamics Thermodynamics
Engineering Thermodynamics Advanced
Thermodynamics for Engineers Thermodynamics for
Engineers Engineering Thermodynamics with Worked
Examples Engineering Thermodynamics Advanced
Thermodynamics for Engineers Thermodynamics for
Engineers Heat Engineering Thermodynamics for
Engineers Applied Thermodynamics for Engineers
Thermodynamics**

***Thermodynamic Approaches in Engineering Systems* responds to the need for a synthesizing volume that throws light upon the extensive field of thermodynamics from a chemical engineering perspective that applies basic ideas and key results from the field to chemical engineering problems. This book outlines and interprets the most valuable achievements in applied non-equilibrium thermodynamics obtained within the recent fifty years. It synthesizes nontrivial achievements of thermodynamics in important branches of chemical and biochemical engineering. Readers will gain an update on what has been achieved, what new research problems could be stated, and what kind of further studies should be developed within specialized research. Presents clearly structured chapters beginning with an introduction, elaboration of the process, and results summarized in a conclusion
Written by a first-class expert in the field of advanced**

methods in thermodynamics Provides a synthesis of recent thermodynamic developments in practical systems Presents very elaborate literature discussions from the past fifty years Enables you to easily advance from thermodynamics principles to applications

Thermodynamics for the Practicing Engineer, as the title suggests, is written for all practicing engineers and anyone studying to become one. Its focus therefore is on applications of thermodynamics, addressing both technical and pragmatic problems in the field. Readers are provided a solid base in thermodynamics theory; however, the text is mostly dedicated to demonstrating how theory is applied to solve real-world problems. This text's four parts enable readers to easily gain a foundation in basic principles and then learn how to apply them in practice: Part One: Introduction. Sets forth the basic principles of thermodynamics, reviewing such topics as units and dimensions, conservation laws, gas laws, and the second law of thermodynamics. Part Two: Enthalpy Effects. Examines sensible, latent, chemical reaction, and mixing enthalpy effects. Part Three: Equilibrium Thermodynamics. Addresses both principles and calculations for phase, vapor-liquid, and chemical reaction equilibrium. Part Four: Other Topics. Reviews such important issues as economics, numerical methods, open-ended problems, environmental concerns, health and safety management, ethics, and exergy. Throughout the text, detailed illustrative examples demonstrate how all the

principles, procedures, and equations are put into practice. Additional practice problems enable readers to solve real-world problems similar to the ones that they will encounter on the job. Readers will gain a solid working knowledge of thermodynamics principles and applications upon successful completion of this text. Moreover, they will be better prepared when approaching/addressing advanced material and more complex problems. Provides an essential treatment of the subject and rigorous methods to solve all kinds of energy engineering problems. Aspiring engineers need a text that prepares them to use thermodynamics in professional practice. Thermodynamics instructors need a concise textbook written for a one-semester undergraduate course—a text that foregoes clutter and unnecessary details but furnishes the essential facts and methods. Thermodynamics for Engineers, Second Edition continues to fill both those needs. Paying special attention to the learning process, the author has developed a unique, practical guide to classical thermodynamics. His approach is remarkably cohesive. For example, he develops the same example through his presentation of the first law and both forms of the second law—entropy and exergy. He also unifies his treatments of the conservation of energy, the creation of entropy, and the destruction of availability by using a balance equation for each, thus emphasizing the commonality between the laws and allowing easier comprehension and use. This Second Edition includes

a new chapter on thermodynamic property relations and gives updated, expanded problem sets in every chapter. Accessible, practical, and cohesive, the text builds a solid foundation for advanced engineering studies and practice. It exposes students to the "big picture" of thermodynamics, and its streamlined presentation allows glimpses into important concepts and methods rarely offered by texts at this level.

What's New in This Edition: Updated and expanded problem sets New chapter on thermodynamic property relations Updated chapter on heat transfer Electronic figures available upon qualifying course adoption End-of-chapter poems to summarize engineering principles This leading text in the field maintains its engaging, readable style while presenting a broader range of applications that motivate engineers to learn the core thermodynamics concepts. Two new coauthors help update the material and integrate engaging, new problems. Throughout the chapters, they focus on the relevance of thermodynamics to modern engineering problems. Many relevant engineering based situations are also presented to help engineers model and solve these problems. Tough Test Questions? Missed Lectures? Not Enough Time? Fortunately for you, there's Schaum's Outlines. More than 40 million students have trusted Schaum's to help them succeed in the classroom and on exams. Schaum's is the key to faster learning and higher grades in every subject. Each Outline presents all the essential course

information in an easy-to-follow, topic-by-topic format. You also get hundreds of examples, solved problems, and practice exercises to test your skills. This Schaum's Outline gives you Practice problems with full explanations that reinforce knowledge Coverage of the most up-to-date developments in your course field In-depth review of practices and applications Fully compatible with your classroom text, Schaum's highlights all the important facts you need to know. Use Schaum's to shorten your study time-and get your best test scores! Schaum's Outlines-Problem Solved. This book is meant as a bridge between physicists, chemists and engineers. It has been written on the basis of many years of teaching at the Chemical Engineering School, Norwegian University of Science and Technology, Trondheim, Norway. This new second edition has been revised in order to put more emphasis on the use of irreversible thermodynamics as a tool for development of sustainable chemical and mechanical processes in the industry. More than 40 million sold in the Schaum's Outline series! This ideal review for the thousands of students who enroll in thermodynamics courses Thermodynamics for Engineers is intended to help engineering students in their understanding of the discipline in a more concise, ordered way than that used in standard textbooks, which are often filled with extraneous material never addressed in the classroom. This edition conforms to the more user-friendly, pragmatic approach now used in most classes. The

outline provides practice sets to allow students to work through the theory they've learned. Material is organized by discrete topics such as gas cycles, vapor cycles, and refrigeration cycles. Practice tests simulate the quizzes and tests given in class. There are also 500 fully solved problems, as well as 180 questions of the type that appear on the engineers' qualifying exam. This new edition boasts problem-solving videos available online and embedded in the ebook version. 500 fully solved problems Problem-solving videos available online and embedded in the ebook version Chapter on refrigeration cycles Nomenclature reflects current usage Four sample tests for the engineering qualifying exam 180 exam-type questions similar to those used on the engineering qualifying exam Helpful material for the following courses: Thermodynamics; Engineering Thermodynamics; Principles of Thermodynamics; Fundamentals of Thermodynamics; Thermodynamics I & II Modern Engineering Thermodynamics is designed for use in a standard two-semester engineering thermodynamics course sequence. The first half of the text contains material suitable for a basic Thermodynamics course taken by engineers from all majors. The second half of the text is suitable for an Applied Thermodynamics course in mechanical engineering programs. The text has numerous features that are unique among engineering textbooks, including historical vignettes, critical thinking boxes, and case studies. All are designed to

bring real engineering applications into a subject that can be somewhat abstract and mathematical. Over 200 worked examples and more than 1,300 end of chapter problems provide opportunities to practice solving problems related to concepts in the text. Provides the reader with clear presentations of the fundamental principles of basic and applied engineering thermodynamics. Helps students develop engineering problem solving skills through the use of structured problem-solving techniques. Introduces the Second Law of Thermodynamics through a basic entropy concept, providing students a more intuitive understanding of this key course topic. Covers Property Values before the First Law of Thermodynamics to ensure students have a firm understanding of property data before using them. Over 200 worked examples and more than 1,300 end of chapter problems offer students extensive opportunity to practice solving problems. Historical Vignettes, Critical Thinking boxes and Case Studies throughout the book help relate abstract concepts to actual engineering applications. For greater instructor flexibility at exam time, thermodynamic tables are provided in a separate accompanying booklet. Available online testing and assessment component helps students assess their knowledge of the topics. Email textbooks@elsevier.com for details. The laws of thermodynamics have wide ranging practical applications in all branches of engineering. This

invaluable textbook covers all the subject matter in a typical undergraduate course in engineering thermodynamics, and uses carefully chosen worked examples and problems to expose students to diverse applications of thermodynamics. This new edition has been revised and updated to include two new chapters on thermodynamic property relations, and the statistical interpretation of entropy. Problems with numerical answers are included at the end of each chapter. As a guide, instructors can use the examples and problems in tutorials, quizzes and examinations. Request Inspection Copy THERMODYNAMICS FOR ENGINEERS focuses on outcome-based learning, which has been identified by ABET as an essential aspect of engineering curricula. Learning outcomes are listed at the start of each chapter and identified as completed at relevant places in the text, followed by a summary at the end of each chapter. Authors Kenneth Kroos and Merle Potter bring decades of teaching experience to a clear writing style that describes key concepts without straying from the course. The language of thermodynamics is explained in careful detail so that students can quickly understand the concepts presented and the analysis techniques used. Extensive use of practical examples demonstrates the proper set-up and solution of problems. These skills are then further developed using a wide variety of homework problems. Some homework problems are presented with an increased degree of complexity to

allow the instructor to challenge the more accomplished. THERMODYNAMICS FOR ENGINEERS focuses on clearly outlining the role of thermodynamics in real engineering. It takes students through clear explanations of concepts, followed by mathematical techniques of analysis and applications of these in solving engineering problems. Important Notice: Media content referenced within the product description or the product text may not be available in the ebook version. Kjelstrup, Bedeaux, Johannessen, and Gross describe what non-equilibrium thermodynamics is in a simple and practical way and how it can add to engineering design. They explain how to describe proper equations of transport that are more precise than those used so far, and how to use them to understand the waste of energy resources in central process units in the industry. The authors introduce the entropy balance as an additional equation to use in engineering; to create consistent thermodynamic models, and to systematically minimize energy losses that are connected with the transport of heat, mass, charge and momentum. Non-equilibrium Thermodynamics for Engineers teaches the essence of non-equilibrium thermodynamics and its applications at a level comprehensible to engineering students, practitioner engineers, and scientists working on industrial problems. The book may be used as a textbook in basic engineering curricula or graduate courses. Excerpt from Heat Engineering: A d104 Book

of Applied Thermodynamics, for Engineers and Students, in Technical Schools Multiple expansion engines action-combined Cards - Computation of Cards for Construction - Equivalent Work done by One Cylinder - Determination of Relative Sizes of Cylinders - Jacketing - Reheaters - Governing -bleeding Engines or Turbines - Regenerative Engines - Testing and Analysis - Binary Engines - Topics - Problems. About the Publisher Forgotten Books publishes hundreds of thousands of rare and classic books. Find more at www.forgottenbooks.com This book is a reproduction of an important historical work. Forgotten Books uses state-of-the-art technology to digitally reconstruct the work, preserving the original format whilst repairing imperfections present in the aged copy. In rare cases, an imperfection in the original, such as a blemish or missing page, may be replicated in our edition. We do, however, repair the vast majority of imperfections successfully; any imperfections that remain are intentionally left to preserve the state of such historical works. The eighth edition of the bestseller Thermodynamics: An Engineering Approach moves students toward a clear understanding and firm grasp of the basic principles of thermodynamics. This textbook communicates directly with tomorrow's engineers in a simple yet precise manner that encourages creative and imaginative thinking and is read by students with interest and enthusiasm all over the world."--Publisher's website This book differs from

other thermodynamics texts in its objective which is to provide engineers with the concepts, tools, and experience needed to solve practical real-world energy problems. The presentation integrates computer tools (e.g., EES) with thermodynamic concepts to allow engineering students and practising engineers to solve problems they would otherwise not be able to solve. The use of examples, solved and explained in detail, and supported with property diagrams that are drawn to scale, is ubiquitous in this textbook. The examples are not trivial, drill problems, but rather complex and timely real world problems that are of interest by themselves. As with the presentation, the solutions to these examples are complete and do not skip steps. Similarly the book includes numerous end of chapter problems, both typeset and online. Most of these problems are more detailed than those found in other thermodynamics textbooks. The supplements include complete solutions to all exercises, software downloads, and additional content on selected topics. These are available at the book web site www.cambridge.org/KleinandNellis. Advanced Thermodynamics Engineering, Second Edition is designed for readers who need to understand and apply the engineering physics of thermodynamic concepts. It employs a self-teaching format that reinforces presentation of critical concepts, mathematical relationships, and equations with concrete physical examples and explanations of

applications—to help readers apply principles to their own real-world problems. Less Mathematical/Theoretical Derivations—More Focus on Practical Application Because both students and professionals must grasp theory almost immediately in this ever-changing electronic era, this book—now completely in decimal outline format—uses a phenomenological approach to problems, making advanced concepts easier to understand. After a decade teaching advanced thermodynamics, the authors infuse their own style and tailor content based on their observations as professional engineers, as well as feedback from their students. Condensing more esoteric material to focus on practical uses for this continuously evolving area of science, this book is filled with revised problems and extensive tables on thermodynamic properties and other useful information. The authors include an abundance of examples, figures, and illustrations to clarify presented ideas, and additional material and software tools are available for download. The result is a powerful, practical instructional tool that gives readers a strong conceptual foundation on which to build a solid, functional understanding of thermodynamics engineering. Building up gradually from first principles, this unique introduction to modern thermodynamics integrates classical, statistical and molecular approaches and is especially designed to support students studying chemical and biochemical

engineering. In addition to covering traditional problems in engineering thermodynamics in the context of biology and materials chemistry, students are also introduced to the thermodynamics of DNA, proteins, polymers and surfaces. It includes over 80 detailed worked examples, covering a broad range of scenarios such as fuel cell efficiency, DNA/protein binding, semiconductor manufacturing and polymer foaming, emphasizing the practical real-world applications of thermodynamic principles; more than 300 carefully tailored homework problems, designed to stretch and extend students' understanding of key topics, accompanied by an online solution manual for instructors; and all the necessary mathematical background, plus resources summarizing commonly used symbols, useful equations of state, microscopic balances for open systems, and links to useful online tools and datasets. Advanced Thermodynamics for Engineers, Second Edition introduces the basic concepts of thermodynamics and applies them to a wide range of technologies. Authors Desmond Winterbone and Ali Turan also include a detailed study of combustion to show how the chemical energy in a fuel is converted into thermal energy and emissions; analyze fuel cells to give an understanding of the direct conversion of chemical energy to electrical power; and provide a study of property relationships to enable more sophisticated analyses to be made of irreversible thermodynamics, allowing for new ways of efficiently

covering energy to power (e.g. solar energy, fuel cells). Worked examples are included in most of the chapters, followed by exercises with solutions. By developing thermodynamics from an explicitly equilibrium perspective and showing how all systems attempt to reach equilibrium (and the effects of these systems when they cannot), Advanced Thermodynamics for Engineers, Second Edition provides unparalleled insight into converting any form of energy into power. The theories and applications of this text are invaluable to students and professional engineers of all disciplines. Includes new chapter that introduces basic terms and concepts for a firm foundation of study Features clear explanations of complex topics and avoids complicated mathematical analysis Updated chapters with recent advances in combustion, fuel cells, and more Solutions manual will be provided for end-of-chapter problems Master the principles of thermodynamics, and understand their practical real-world applications, with this deep and intuitive undergraduate textbook. Excerpt from Heat Engineering: A Text Book of Applied Thermodynamics, for Engineers and Students, in Technical Schools For many years the author has given lectures supplementing the text-books used as a basis for a course in heat engineering. His aim in preparing this book has been to bring together his various notes with statements of the investigations and writings of others to make a complete treatment of the important phases

of this subject. In doing this he has given credit to the authors and investigators quoted. Certain of the original sources have been quoted so that the student may learn the use of references. It is hoped that many studying this book will refer to these original papers. The work presupposes a course in theoretical thermodynamics such as that given in the treatises of Wood, Peabody or Goodenough. Because of the difference in symbols, nomenclature or point of view of various authors and to serve for reference or for the derivation of formulæ used in the text, the first chapter of this book has been written. It is not intended that this chapter shall be used as a part of the course for it is an outline only of the thermodynamic theory. It should be used to give a review of the subject or as a basis for the formulæ used. In shaping this chapter the author has been guided by his experience in teaching this subject from many texts. The treatment of availability and entropy has been based on the excellent work on thermodynamics by Goodenough. Numerical problems have been solved at various points in the text to illustrate the principles of the subject and to apply them to actual engineering work: The problems have been solved in detail to give the student one manner of attack as well as an order for the arrangement of computations for clearness. Unless the student can apply the various formulæ and theories he has failed to attain that for which this book was written. In addition to the problems and solutions a series of

questions on the various topics of the text and a set of problems illustrating their use have been placed at the end of each chapter. These may be used by the student in preparation of an assignment or by the teacher for blackboard recitations. About the Publisher Forgotten Books publishes hundreds of thousands of rare and classic books. Find more at www.forgottenbooks.com This book is a reproduction of an important historical work. Forgotten Books uses state-of-the-art technology to digitally reconstruct the work, preserving the original format whilst repairing imperfections present in the aged copy. In rare cases, an imperfection in the original, such as a blemish or missing page, may be replicated in our edition. We do, however, repair the vast majority of imperfections successfully; any imperfections that remain are intentionally left to preserve the state of such historical works. Considered as particularly difficult by generations of students and engineers, thermodynamics applied to energy systems can now be taught with an original instruction method. Energy Systems applies a completely different approach to the calculation, application and theory of multiple energy conversion technologies. It aims to create the reader's foundation for understanding and applying the design principles to all kinds of energy cycles, including renewable energy. Proven to be simpler and more reflective than existing methods, it deals with energy system modeling, instead of the thermodynamic foundations, as the primary objective.

Although its style is drastically different from other textbooks, no concession is done to coverage: with encouraging pace, the complete range from basic thermodynamics to the most advanced energy systems is addressed. The accompanying ThermoOptim™ portal (http://direns.mines-paristech.fr/Sites/Thopt/en/co/_Arborescence_web.html) presents the software and manuals (in English and French) to solve over 200 examples, and programming and design tools for exercises of all levels of complexity. The reader is explained how to build appropriate models to bridge the technological reality with the theoretical basis of energy engineering. Offering quick overviews through e-learning modules moreover, the portal is user-friendly and enables to quickly become fully operational. Students can freely download the ThermoOptim™ modeling software demo version (in seven languages) and extended options are available to lecturers. A professional edition is also available and has been adopted by many companies and research institutes worldwide - www.thermooptim.org This volume is intended as for courses in applied thermodynamics, energy systems, energy conversion, thermal engineering to senior undergraduate and graduate-level students in mechanical, energy, chemical and petroleum engineering. Students should already have taken a first year course in thermodynamics. The refreshing approach and exceptionally rich coverage make it a great reference

tool for researchers and professionals also. Contains International Units (SI). This textbook comprehensively covers the fundamentals and advanced concepts of thermodynamics in a single volume. It provides a detailed discussion of advanced concepts that include energy efficiency, energy sustainability, energy security, organic Rankine cycle, combined cycle power plants, combined cycle power plant integrated with organic Rankine cycle and absorption refrigeration system, integrated coal gasification combined cycle power plants, energy conservation in domestic refrigerators, and next-generation low-global warming potential refrigerants. Pedagogical features include solved problems and unsolved exercises interspersed throughout the text for better understanding. This textbook is primarily written for senior undergraduate students in the fields of mechanical, automobile, chemical, civil, and aerospace engineering for courses on engineering thermodynamics/thermodynamics and for graduate students in thermal engineering and energy engineering for courses on advanced thermodynamics. It is accompanied by teaching resources, including a solutions manual for instructors.

FEATURES Provides design and experimental problems for better understanding
Comprehensively discusses power cycles and refrigeration cycles and their advancements
Explores the design of energy-efficient buildings to reduce energy consumption
Property tables, charts, and multiple-choice questions

comprise appendices of the book and are available at <https://www.routledge.com/9780367646288>. Energy-its discovery, its availability, its use-concerns all of us in general and the engineers of today and tomorrow in particular. The study of thermodynamics-the science of energy-is a critical element in the education of all types of engineers. Engineering Thermodynamics provides a thorough introduction to the art and science of engineering thermodynamics. It describes in a straightforward fashion the basic tools necessary to obtain quantitative solutions to common engineering applications involving energy and its conversion, conservation, and transfer. This book is directed toward sophomore, junior, and senior students who have studied elementary physics and calculus and who are majoring in mechanical engineering; it serves as a convenient reference for other engineering disciplines as well. The first part of the book is devoted to basic thermodynamic principles, essentially presented in the classic way; the second part applies these principles to many situations, including air conditioning and the interpretation of statistical phenomena. Mechanical Engineering Thermodynamics is a subject that all engineering students have to face and that most of them treat with great respect. This makes it all the more important to offer a good and easy-to-understand approach to the laws of energy conversion. This is what this textbook is intended to do: It covers the basics of classical technical thermodynamics as they

are typically taught at universities: The first and second law of thermodynamics as well as equations of state are explained for idealized and real fluids which are subject to a phase change. Thermodynamic mixtures, e.g. humid air, are treated as well as chemical reactions. Components and thermodynamic cycle that convert energy are presented. The book attaches great importance to drawings and illustrations, which should make it easier to comprehend complex matter. Technical applications and apparatus are presented and explained. Numerous exercises and examples conclude the book and contribute to a better understanding of the theory. Clearly connects macroscopic and microscopic thermodynamics and explains non-equilibrium behavior in kinetic theory and chemical kinetics. Although the basic theories of thermodynamics are adequately covered by a number of existing texts, there is little literature that addresses more advanced topics. In this comprehensive work the author redresses this balance, drawing on his twenty-five years of experience of teaching thermodynamics at undergraduate and postgraduate level, to produce a definitive text to cover thoroughly, advanced syllabuses. The book introduces the basic concepts which apply over the whole range of new technologies, considering: a new approach to cycles, enabling their irreversibility to be taken into account; a detailed study of combustion to show how the chemical energy in a fuel is converted into thermal energy and emissions; an

analysis of fuel cells to give an understanding of the direct conversion of chemical energy to electrical power; a detailed study of property relationships to enable more sophisticated analyses to be made of both high and low temperature plant and irreversible thermodynamics, whose principles might hold a key to new ways of efficiently covering energy to power (e.g. solar energy, fuel cells). Worked examples are included in most of the chapters, followed by exercises with solutions. By developing thermodynamics from an explicitly equilibrium perspective, showing how all systems attempt to reach a state of equilibrium, and the effects of these systems when they cannot, the result is an unparalleled insight into the more advanced considerations when converting any form of energy into power, that will prove invaluable to students and professional engineers of all disciplines. Applied Thermodynamics for Engineering Technologists provides a complete introduction to the principles of thermodynamics for degree level students on courses in mechanical, aeronautical, chemical, environmental and energy engineering and engineering science courses. The fifth edition of this classic text for applied courses has been completely revised and updated to take account of modern teaching methods and perspectives, with the emphasis placed on the application of theory to real processes and plant. New for this edition is a section on energy recovery, including pinch technology and a discussion of the

thinning of the ozone layer due to the use of CFCs. Examples and problems using the refrigerant 134A replace the previous references to CFC R12. In addition, the discussion of energy sources, their uses and management, has been expanded and improved and there is now extensive coverage of the combined heat and power section. The material on turbines, compressors, nozzles and propulsion is presented in a more logical sequence but retains important information on the differences between gas and steam turbines. Finally the section on refrigeration gives more prominence to the heat pump and vapour absorption plant. This textbook takes an interdisciplinary approach to the subject of thermodynamics and is therefore suitable for undergraduates in chemistry, physics and engineering courses. The book is an introduction to phenomenological thermodynamics and its applications to phase transitions and chemical reactions, with some references to statistical mechanics. It strikes the balance between the rigorousness of the Callen text and phenomenological approach of the Atkins text. The book is divided in three parts. The first introduces the postulates and laws of thermodynamics and complements these initial explanations with practical examples. The second part is devoted to applications of thermodynamics to phase transitions in pure substances and mixtures. The third part covers thermodynamic systems in which chemical reactions take place. There are some sections on more

advanced topics such as thermodynamic potentials, natural variables, non-ideal mixtures and electrochemical reactions, which make this book of suitable also to post-graduate students. This book is a beginners introduction to chemical thermodynamics for engineers. In the textbook efforts have been made to visualize as clearly as possible the main concepts of thermodynamic quantities such as enthalpy and entropy, thus making them more perceivable. Furthermore, intricate formulae in thermodynamics have been discussed as functionally unified sets of formulae to understand their meaning rather than to mathematically derive them in detail. In this textbook, the affinity of irreversible processes, defined by the second law of thermodynamics, has been treated as the main subject, rather than the equilibrium of chemical reactions. The concept of affinity is applicable in general not only to the processes of chemical reactions but also to all kinds of irreversible processes. This textbook also includes electrochemical thermodynamics in which, instead of the classical phenomenological approach, molecular science provides an advanced understanding of the reactions of charged particles such as ions and electrons at the electrodes. Recently, engineering thermodynamics has introduced a new thermodynamic potential called exergy, which essentially is related to the concept of the affinity of irreversible processes. This textbook discusses the relation between exergy and affinity and

explains the exergy balance diagram and exergy vector diagram applicable to exergy analyses in chemical manufacturing processes. This textbook is written in the hope that the readers understand in a broad way the fundamental concepts of energy and exergy from chemical thermodynamics in practical applications. Finishing this book, the readers may easily step forward further into an advanced text of their specified line. - Visualizes the main concepts of thermodynamics to show the meaning of the quantities and formulae. - Focuses mainly on the affinity of irreversible processes and the related concept of exergy. - Provides an advanced understanding of electrochemical thermodynamics. Furthermore, a chapter on the microscopic implications of the entropy function and the second law is also included. The book describes in a simple and practical way what non-equilibrium thermodynamics is and how it can add to engineering fields. It explains how to describe proper equations of transport, more precise than used so far, and how to use them to understand the waste of energy resources in central unit processes in the industry. It introduces the entropy balance as an additional equation to use, to create consistent thermodynamic models, and a systematic method for minimizing energy losses that are connected with transport of heat, mass, charge, momentum and chemical reactions. Aspiring engineers need a text that prepares them to use thermodynamics in professional practice. Thermodynamics instructors

need a concise textbook written for a one-semester undergraduate course—a text that foregoes clutter and unnecessary details but furnishes the essential facts and methods. Thermodynamics for Engineers, Second Edition continues to fill both those needs. Paying special attention to the learning process, the author has developed a unique, practical guide to classical thermodynamics. His approach is remarkably cohesive. For example, he develops the same example through his presentation of the first law and both forms of the second law—entropy and exergy. He also unifies his treatments of the conservation of energy, the creation of entropy, and the destruction of availability by using a balance equation for each, thus emphasizing the commonality between the laws and allowing easier comprehension and use. This Second Edition includes a new chapter on thermodynamic property relations and gives updated, expanded problem sets in every chapter. Accessible, practical, and cohesive, the text builds a solid foundation for advanced engineering studies and practice. It exposes students to the "big picture" of thermodynamics, and its streamlined presentation allows glimpses into important concepts and methods rarely offered by texts at this level.

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